$\left.\mathrm{OCH}_{3}\right), 4.8(1 \mathrm{H}, \mathrm{m}$, olefin), $7.2-8(5 \mathrm{H}, \mathrm{m}$, arom. $)$; mass spectrum, $m / e$ $240\left(\mathrm{M}^{+}, 100\right), 238$ (35). Precise mass determination: calcd for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{O} 240.115022$; found 240.1150. Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{O}: \mathrm{C}$, 79.97 ; H, 6.71; O, 13.32. Found: C, 79.59; H, 6.70; O, 13.63.

1,4-Dihydro-1,8-dimethylphenanthrene (3d): mp $33-34^{\circ} \mathrm{C}$; NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.38\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3}\right), 2.65\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3}\right), 3.65(3 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{2}$ ), 5.9 ( $2 \mathrm{H}, \mathrm{m}$, olefin), $7.5(5 \mathrm{H}, \mathrm{m}$, arom.) ; mass spectrum $\mathrm{m} / \mathrm{e} 208$ $\left(\mathrm{M}^{+}, 100\right), 206$ (70). Precise mass determination: calcd for $\mathrm{C}_{16} \mathrm{H}_{16}$ 208.125194; found 208.1251. Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{26}$ : C, $92.26 ; \mathrm{H}, 7.74$. Found: C, 92.32; H, 7.68.
Photolysis of 2-Vinylbiphenyl (7). A solution of compound 7 $\left(10^{-2} \mathrm{M}\right)$ in propylamine was purged with nitrogen and irradiated with Rul $3000-\AA$ lamps for 8 h . After the amine was removed under reduced pressure, the crude photoreaction product was analyzed by NMR and purified by column chromatography on alumina.

From 180 mg of $7,125 \mathrm{mg}$ of $6 \mathrm{a}(70 \%)$ and 15 mg of $9(8 \%)$ were obtained. The structures of these compounds were established by comparison with authentic samples.

2-Ethylbiphenyl (9) was prepared by reduction of the starting material in the usual manner.

Acknowledgments. The authors thank Dr. B. M. Carden for linguistic criticism of the manuscript.

Registry No.-la, 103-30-0; 1b, 18869-29-9; 1c, 15638-14-9; 1d, ;36888-18-3; 2a, 645-49-8; 2b, 2510-76-1; 2c, 2510-75-0; 2d, 20657-42-5; 3a, 20244-28-4, 3b, 69795-78-4; 3c, 69795-79-5; 3d, 69795-80-8; 4a, 85-01-8; 4b, 1576-67-6; 4c, 15638-08-1; 4d, 7372-87-4; 5а, 103-29-7; 5b, 538-39-6; 5c, 1657-55-2; $\mathbf{5 d}$, 952-80-7; 6а, 776-35-2; 6b, 69795-81-9; 6c, 69832-49-1: 6d, 69795-82-0; 7, 1587-22-0; 9, 1812-51-7; p, p'-dimethylbenzoin, 1218-89-9; 0,o'-dimethylbenzoin, 4389-39-3; $p$-tolualdehyde, 104-87-0; o-tolualdehyde, 529-20-4; 2-iodobiphenyl, 2113-51-1; acetaldehyde. 75-07-0; 1-biphenylethanol, 16927-84-7.

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## Electrolytic Decarboxylation Reactions. 4. Electrosyntheses of 3-Alkyl-2-cycloalken-1-ol Acetates from 1-Alkyl-2-cycloalkene-1-carboxylic Acids. Preparation of $d \boldsymbol{l}$-Muscone from Cyclopentadecanone

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Received December 1, 1978
Based on stimulating results on the electrolytic acetoxylation of aliphatic carboxylic acids, ${ }^{1}$ we have developed an electrosynthetic procedure for 3-alkyl-2-cycloalken-1-ol acetates (2a) from 1-alkyl-2-cycloalkene-1-carboxylic acids (1c) prepared from alicyclic 2 -oxoalkanoates (3a). Synthetic application of such non-Kolbe type reactions on 3-alkenoates has been paid little attention. ${ }^{2}$ The present 3 -alkyl-2-alken1 -ol synthesis involves a regiospecific acetoxylation at the $\gamma$ position of the acids $1 \mathbf{c}$, which serves as an introducing method for a methyl group at the $\beta$ position of cyclopentadecanone, ${ }^{3}$ leading to dl -muscone.

The 3 -alkenoic acids $1 \mathbf{c}$ were all prepared by (i) alkylation of $3 a$, (ii) reduction of $3 b$ with sodium borohydride or lithium tri-tert-butoxyaluminum hydride, (iii) dehydration of the alcohol 4 a via the corresponding mesylate $\mathbf{4 b}$, and (iv) hy-

drolysis of la in $\sim 70 \%$ overall yields. The electroacetoxylation of $1 \mathbf{c}\left(n=12, \mathrm{R}^{\mathrm{I}}=\mathrm{Me}\right.$ ) was carried out in $\mathrm{AcOH}-t-\mathrm{BuOH}-$ $\mathrm{Et}_{3} \mathrm{~N}$ using platinum electrodes at a constant applied voltage of $\sim 30 \mathrm{~V}\left(36-54 \mathrm{~mA} / \mathrm{cm}^{2}, 153 \mathrm{~F} / \mathrm{mol}\right)$ at $19-22^{\circ} \mathrm{C}$ for 4 h . The electrolysis conditions and results of the related compounds lc are shown in Table I.

Electrolytic decarboxylation of the acids $\mathbf{1 c}$ by loss of two electrons on the anode would provide the tertiary carbonium ion a and subsequent three-carbon anionotropic rearrangement ${ }^{4}$ of the cation a into the secondary carbonium ion $b$. The results (Table I) reveal that electrodecarboxylation of $1 \mathbf{c}$ in

Table I. Conditions ${ }^{\text {a }}$ and Results of Electrolytic Acetoxylation of 1-Alkyl-2-cycloalkene-1-carboxylic Acids 1c

|  | substrate lc |  | mmol | current density, <br> $\mathrm{mA} / \mathrm{cm}^{2}$ | electricity, <br> $\mathrm{F} / \mathrm{mol}$ | time, h |
| ---: | :--- | :---: | :---: | :---: | :---: | :---: |

${ }^{a}$ Electrolyzed at $19-22^{\circ} \mathrm{C}, \mathrm{Pt}\left(3 \mathrm{~cm}^{2}\right)$, in a mixed solution of $\mathrm{AcOH}(1.5 \mathrm{~mL})$, $\mathrm{AcOEt}(3.5 \mathrm{~mL})$, and $t-\mathrm{BuOH}(0.19 \mathrm{~mL})$ using Et $\mathrm{N}_{3} \mathrm{~N}$ $(0.6 \mathrm{~mL})$ as an additive. ${ }^{6}$ Adjusted at 30 V (applied voltage). ${ }^{\text {c }}$ Adjusted at 20 V (applied voltage). ${ }^{d}$ Based on isolated product.

Table II. Conditions ${ }^{a}$ and Results of Electrolytic Methoxylation of 1-Alkyl-2-cycloalkene-1-carboxylic Acids 1c

| substrate 1c |  |  | $\begin{gathered} \text { current density, }{ }^{b} \mathrm{~mA} / \mathrm{cm}^{2} \end{gathered}$ | electricity, $\mathrm{F} / \mathrm{mol}$ | Time, h | $\begin{gathered} \text { yield of } 2 \mathbf{c}+5, \\ \%(\text { ratio }) \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $n$ | $\mathrm{R}^{1}$ | $\overline{\mathrm{mmol}}$ |  |  |  |  |
| 12 | Me | 0.104 | 41-44 | 124 | 3 | $83(1: 2)^{\text {d }}$ |
| 9 | Me | 0.104 | 37-44 | 115 | 3 | $83(1: 2)^{e}$ |
| 3 | $\mathrm{CH}_{3}\left(\mathrm{CH}_{2}\right)_{4}$ | 0.392 | 30-46 | 35 | 3.5 | 57 (7:3) $f$ |

${ }^{a}$ The electrolyses were carried out at $16-26^{\circ} \mathrm{C}, \mathrm{Pt}$ anode ( $3 \mathrm{~cm}^{2}$ ) and SUS-27 stainless steel cathode ( $11 \mathrm{~cm}^{2}$ ), in MeOH ( 15 mL ) with $\mathrm{Et}_{3} \mathrm{~N}(210 \mathrm{mg})$ as an additive. ${ }^{b}$ Applied voltage was adjusted at $20 \mathrm{~V} .{ }^{c}$ Based on isolated products. ${ }^{d}$ Estimated by comparison with ${ }^{1} \mathrm{H}$ NMR signals at $\delta 3.16$ and $3.22\left(\mathrm{~s}, \mathrm{OCH}_{3}\right) .{ }^{e}$ Estimated by comparison with ${ }^{1} \mathrm{H}$ NMR signals at $\delta 3.19$ and 3.26 ( $\mathrm{s}, \mathrm{OCH}_{3}$ ). $f$ Estimated by VPC (silicon GE-30 $10,4 \mathrm{~m} \times 4 \mathrm{~mm}, \mathrm{H}_{2}$ flow rate $35 \mathrm{~mL} / \mathrm{min}$ ): $R_{\mathrm{t}}(\mathrm{min}) 5$ (8.5) and $2 \mathrm{c}(10.2)$.
$\mathrm{AcOH}-t-\mathrm{BuOH}-\mathrm{Et}_{3} \mathrm{~N}$ affords exclusively the acetates 2a via the carbonium ion b. In contrast, electromethoxylation of 1 c ( $n=12, \mathrm{R}^{1}=\mathrm{Me}$ ) conducted in the more polar solvent $\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N}$ using the same apparatus fitted with a platinum anode and a stainless steel (SUS-27) cathode at a constant applied voltage of $\sim 20 \mathrm{~V}\left(41-44 \mathrm{~mA} / \mathrm{cm}^{2}, 124 \mathrm{~F} / \mathrm{mol}\right)$ at $20-26$ ${ }^{\circ} \mathrm{C}$ for 3 h gave a mixture of $2 \mathrm{c}\left(n=12, \mathrm{R}^{1}=\mathrm{Me} ; 28 \%\right.$ yield) and $\mathbf{5}$ ( $n=12, \mathrm{R}^{1}=\mathrm{Me} ; 55 \%$ yield). These results (Table II)

suggest that the strong nucleophilicity of methanol, compared to acetic acid, would preferentially lead to combination with the initially produced cation a to give 5 , whereas the carbonium ion $b$ would be stabilized in the weak nucleophilic medium. ${ }^{5}$
On the other hand, electrodecarboxylation of 8,8a-epoxy-2,2-(ethylenedioxy)decalin-4a-carboxylic acid (6) in $\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N}$ at $20 \mathrm{~V}\left(23-32 \mathrm{~mA} / \mathrm{cm}^{2}\right)$ afforded smoothly the



corresponding enone 7 in $84 \%$ yield as the sole product. However, under the same reaction conditions electrodecarboxylation of the octalin derivative 8 in $\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N}$ provided the mixed products $9(28.6 \%)$ and $10(39.7 \%)$.

Part of our synthetic interest was directed to the conversion of the acetate $\mathbf{2 a}\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right)$ into dl -muscone. After hydrolysis of $\mathbf{2 a}\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right)$ with potassium hydroxide in aqueous methanol, oxidation of $2 b\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right)$ with pyridinium chlorochromate to give 11 and subsequent hydrogenation using palladium on charcoal afforded a $59 \%$ overall yield of $d l$-muscone.


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## Experimental Section

Melting points and boiling points are uncorrected. IR spectra were determined with a JASCO IRA-1 grating spectrometer. ${ }^{1} \mathrm{H}$ NMR spectra were determined at 60 MHz with a Hitachi R-24 or at 100 MHz with a JEOL MH-100 spectrometer. ${ }^{19} \mathrm{C}$ NMR spectra were determined at 25.05 MHz with a JEOL pulsed Fourier transform spectrometer, Model FX-100. Samples were dissolved in $\mathrm{CDCl}_{3}$, and the chemical shift values were expressed in $\delta$ values ( ppm ) relative to $\mathrm{Me}_{4} \mathrm{Si}$ as an internal standard. Elemental analyses were performed in our laboratory.

Methyl 1-Methyl-2-oxocyclopentadecane-1-carboxylate (3b, $\left.n=12, \mathbf{R}^{1}=\mathbf{R}^{2}=\mathbf{M e}\right)$. A mixture of $3 \mathbf{a}^{6}\left(n=12, \mathrm{R}^{2}=\mathrm{Me} ; 425 \mathrm{mg}\right.$, 1.51 mmol ), $\mathrm{K}_{2} \mathrm{CO}_{3}(1.64 \mathrm{~g}, 11.9 \mathrm{mmol})$, and MeI ( $460 \mathrm{mg}, 3.29 \mathrm{mmol}$ ) in acetone ( 12 mL ) was refluxed for 12 h . The insoluble material was separated by centrifugation, and the organic layer was concentrated. The residue was chromatographed ( $\mathrm{SiO}_{2}, 7: 1$ hexane- AcOEt ) to give $386 \mathrm{mg}(86 \%)$ of $3 \mathbf{b}\left(n=12, \mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{Me}\right)$ as a waxy oil: bp $104.0-$ $108.0^{\circ} \mathrm{C}(0.005 \mathrm{~mm}$, Kugelrohr); IR (neat) 1743 (ester $\mathrm{C}=0$ ), 1711 $(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $(60 \mathrm{MHz}) \delta 1.30\left(\mathrm{~s}, 3, \mathrm{CH}_{3}\right), 1.33$ (br s, 24, $\left.\mathrm{CH}_{2}\right), 2.40\left(\mathrm{~m}, 2, \mathrm{COCH}_{2}\right), 3.71\left(\mathrm{~s}, 3, \mathrm{OCH}_{3}\right) ;{ }^{13} \mathrm{C}$ NMR $\delta 18.8(\mathrm{q}$, $\left.\mathrm{CCH}_{3}\right), 22.5(\mathrm{t}), 25.8(\mathrm{t}), 26.1(\mathrm{t}), 26.3(\mathrm{t}), 26.8(\mathrm{t}), 27.0(\mathrm{t}), 27.5(\mathrm{t}), 34.9$ $(\mathrm{t}, \mathrm{C}-15), 37.5(\mathrm{t}, \mathrm{C}-3), 52.2\left(\mathrm{q}, \mathrm{OCH}_{3}\right), 59.4(\mathrm{~s}, \mathrm{C}-1), 174.1(\mathrm{~s}$, ester $\mathrm{C}=\mathrm{O}$ ), 207.7 (s, C-2). Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{32} \mathrm{O}_{3}$ : C. $72.93 ; \mathrm{H}, 10.88$. Found: C, 73.01; H, 10.94 .

Methyl 1-methyl-2-oxocyclododecane-1-carboxylate (3b, $n$ $=9, \mathbf{R}^{1}=R^{2}=\mathbf{M e}$ ) was prepared in the same manner as described above in $90 \%$ yield by treatment of $3 \mathbf{a}^{7}\left(n=9, \mathrm{R}^{2}=\right.$ Me) with MeI: bp $97.0-99.0^{\circ} \mathrm{C}(0.002 \mathrm{~mm}$, Kugelrohr); IR (neat) 1738 (ester $\mathrm{C}=\mathrm{O}$ ), 1710 $(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}(100 \mathrm{MHz}) \delta 0.90,0.97,1.55\left(\mathrm{~s}, 3, \mathrm{CH}_{3}\right), 1.31$
(br s, 18, $\mathrm{CH}_{2}$ ), 2.75-3.26 (m, 2, $\mathrm{COCH}_{2}$ ), 3.71 (s, 3, $\mathrm{OCH}_{3}$ ). Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{26} \mathrm{O}_{3}$ : $\mathrm{C}, 70.83 ; \mathrm{H}, 10.30$. Found: C, $70.92 ; \mathrm{H}, 10.59$.
Methyl 1-pentyl-2-oxocycloheptane-1-carboxylate (3b, $\boldsymbol{n}=$ 4, $\mathbf{R}^{1}=\boldsymbol{n}-\mathrm{C}_{5} \mathbf{H}_{11}, \mathbf{R}^{\mathbf{2}}=\mathbf{M e}$ ) was prepared in the same manner as described above in $87 \%$ yield by the reaction of $3 \mathrm{a}^{7}\left(n=4, \mathrm{R}^{2}=\mathrm{Me}\right.$ ) with pentyl bromide in the presence of KI: bp $55.0-58.5^{\circ} \mathrm{C}(0.017 \mathrm{~mm}$, Kugelrohr); IR (neat) 1738 (ester $\mathrm{C}=0$ ), $1714(\mathrm{C}=0) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $(100 \mathrm{MHz}) \delta 0.89\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.30\left(\mathrm{br} \mathrm{s}, 8, \mathrm{CH}_{2}\right), 1.50-2.30$ $\left(\mathrm{m}, 8, \mathrm{CH}_{2}\right), 2.39-2.76\left(\mathrm{~m}, 2, \mathrm{COCH}_{2}\right), 3.78\left(\mathrm{~s}, 3, \mathrm{OCH}_{3}\right)$. Anal. Caled for $\mathrm{C}_{14} \mathrm{H}_{24} \mathrm{O}_{3}: \mathrm{C}, 69.96 ; \mathrm{H}, 10.07$. Found: C, 70.08 ; $\mathrm{H}, 10.25$.
Methyl 1-hexyl-2-oxocyclopentane-1-carboxylate (3b, $n=2$, $\mathbf{R}^{\mathbf{1}}=\boldsymbol{n}-\mathbf{C}_{6} \mathbf{H}_{13}, \mathbf{R}^{2}=\mathbf{M e}$ ) was prepared in the same manner as de scribed above in $93 \%$ yield by the reaction of $3 \mathbf{a}^{8}\left(n=2, \mathrm{R}^{2}=\mathrm{Me}\right.$ ) with hexyl bromide in the presence of KI: bp $90.0-93.0^{\circ} \mathrm{C}(2 \mathrm{~mm}$, Kugelrohr); IR (neat) $1755\left(\mathrm{C}=0\right.$ ), 1730 (ester $\mathrm{C}=0$ ) $\mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 60 $\mathrm{MHz}) \delta 0.86\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.28\left(\mathrm{br} \mathrm{s}, 8, \mathrm{CH}_{2}\right), 1.35-2.70(\mathrm{~m}, 8$, $\mathrm{CH}_{2}$ ), 3.68 ( $\mathrm{s}, 3, \mathrm{OCH}_{3}$ ). Anal. Caled for $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{3}$ : $\mathrm{C}, 68.99 ; \mathrm{H}, 9.80$. Found: C, $68.80 ; \mathrm{H}, 9.80$.
Methyl 2-Hydroxy-1-methylcyclopentadecane-1-carboxylate ( $\mathbf{4 a}, \boldsymbol{n}=12, \mathbf{R}^{1}=\mathbf{R}^{\mathbf{2}}=\mathbf{M e}$ ). To a cold solution ( $0-5^{\circ} \mathrm{C}$ ) of $\mathbf{3 b}(n=12$, $\left.\mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{Me} ; 106 \mathrm{mg}, 0.36 \mathrm{mmol}\right)$ in $\mathrm{MeOH}(2 \mathrm{~mL})$ was added dropwise a solution of $\mathrm{NaBH}_{4}(48 \mathrm{mg}, 1.27 \mathrm{mmol})$ in water $(0.5 \mathrm{~mL})$, The mixture was stirred for 3 h at room temperature, quenched with cold aqueous $10 \% \mathrm{AcOH}$, and worked up in the usual manner to give $96 \mathrm{mg}(89 \%)$ of $\mathbf{4 a}\left(n=12, \mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{Me}\right)$ as an oil: bp $112.0-114.5^{\circ} \mathrm{C}$ 0.004 mm , Kugelrohr); IR (neat) 3500 (OH), 1730 (shoulder), 1714 (ester $\mathrm{C}=0$ ) $\mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 60 MHz ) $\delta 1.13\left(\mathrm{~s}, 3, \mathrm{CH}_{3}\right), 1.33(\mathrm{brs}$, $26, \mathrm{CH}_{2}$ ), 2.80 (br s, 1, OH), 3.69 ( $\mathrm{s}, 3, \mathrm{OCH}_{3}$ ), 3.35-4.10 (m, 1, CHO). Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{34} \mathrm{O}_{3}: \mathrm{C}, 72.44 ; \mathrm{H}, 11.48$. Found: $\mathrm{C}, 72.54 ; \mathrm{H}$, 11.49.

Methyl 2-hydroxy-1-methylcyclododecane-1-carboxylate (4a, $n=\mathbf{9 ,} \mathbf{R}^{\mathbf{1}}=\mathbf{R}^{\mathbf{2}}=\mathbf{M e}$ ) was prepared in the same manner as described above in $96 \%$ yield: mp $115.5-116.5^{\circ} \mathrm{C}$; IR (Nujol) $3510(\mathrm{OH}), 1710$ (ester $\mathrm{C}=0$ ) $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $(100 \mathrm{MHz}) \delta 1.17\left(\mathrm{~s}, 3, \mathrm{CH}_{3}\right), 1.36$ (br s, $20, \mathrm{CH}_{2}$ ), 2.88 (br, $\left.1, \mathrm{OH}\right), 3.72$ ( $\mathrm{s}, 3, \mathrm{OCH}_{3}$ ), $4.22(\mathrm{~d}, 1, J=9 \mathrm{~Hz}, \mathrm{CHO}$ ). Anal. Caled for $\mathrm{C}_{15} \mathrm{H}_{28} \mathrm{O}_{3}$ : C, 70.27: H, 11.01. Found: C, 70.15; H, 11.02.

Methyl 2-hydroxy-1-pentylcycloheptane-1-carboxylate (4a, $\boldsymbol{n}=\mathbf{4}, \mathrm{R}^{\mathbf{1}}=\boldsymbol{n}-\mathrm{C}_{5} \mathbf{H}_{11}, \mathrm{R}^{\mathbf{2}}=\mathbf{M e}$ ) was prepared in the same manner as described above in $90 \%$ yield: bp $96.0-98.0^{\circ} \mathrm{C}(0.03 \mathrm{~mm}$, Kugelrohr); IR (neat) 350010 H ), 1727 (ester $\mathrm{C}=0$ ), 1710 (shoulder) $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 0.88\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right.$ ), 1.26 (br, $8, \mathrm{CH}_{2}$ ), 1.4()$-2.25\left(\mathrm{~m}, 10, \mathrm{CH}_{2}\right.$ ), 2.59 (br, $\left.1, \mathrm{OH}\right), 3.70\left(\mathrm{~s}, 3, \mathrm{OCH}_{3}\right), 3.84-4.16$ (m, 1, CHO). Anal. Calced for $\mathrm{C}_{14} \mathrm{H}_{26} \mathrm{O}_{3}: \mathrm{C}, 69.38 ; \mathrm{H}, 10.81$. Found: C , 69.48; H, 10.86.

Ethyl 2-hydroxy-1-pentylcyclohexane-1-carboxylate (4a, $\boldsymbol{n}$ $=3, \mathbf{R}^{1}=n-\mathbf{C}_{5} \mathbf{H}_{11}, \mathbf{R}^{2}=\mathbf{E t}$ ) was prepared in the same manner in $94 \%$ yield by the reaction of $3 \mathrm{~b}^{9}\left(n=3, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}, \mathrm{R}^{2}=\mathrm{Et}\right)$ and $\mathrm{NaBH}_{4}$ : bp $77.5-79.0^{\circ} \mathrm{C}(0.006 \mathrm{~mm}$, Kugelrohr); IR (neat) $3500(\mathrm{OH})$, 1725. 1701 (ester $\mathrm{C}=0$ ) $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 0.88(\mathrm{t}, 3 . J=$ $\left.6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.00-2.30\left(\mathrm{~m}, 16, \mathrm{CH}_{2}\right), 1.26,1.29\left(\mathrm{t}, 3, \mathrm{~J}=7 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 3.15$ (b), $1, \mathrm{OH}$ ), 3.48, 3.88 (dd, $1, J=8$ and $4 \mathrm{~Hz}, \mathrm{CHO}$ ), $4.02,4.04(\mathrm{q}, 2$. $J=7 \mathrm{~Hz}_{2}, \mathrm{CH}_{2} \mathrm{O}$ ). Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{26} \mathrm{O}_{3}: \mathrm{C}, 69.38 ; \mathrm{H}, 10.81$. Found C. $69.47 ; \mathrm{H}, 10.96$.

Methyl 1-hexyl-2-hydroxycyclopentane-1-carboxylate (4a, $\boldsymbol{n}=\mathbf{2}, \mathbf{R}^{1}=\boldsymbol{n}-\mathrm{C}_{6} \mathbf{H}_{13}, \mathbf{R}^{\mathbf{2}}=\mathbf{M e}$ ) was prepared in $96 \%$ yield by the reaction of 3 b ( $\left.n=2 . \mathrm{R}^{1}=n-\mathrm{C}_{6} \mathrm{H}_{1 ; 3}, \mathrm{R}^{2}=\mathrm{Me}\right)$ and $\mathrm{LiAl}(t-\mathrm{BuO}) ; \mathrm{H}:$ bp 59.0-61.0 ${ }^{\circ} \mathrm{C}(0.003 \mathrm{~mm}$. Kugelrohr): IR (neat) $3450(\mathrm{OH}), 1725$. 1715 lester $\mathrm{C}=0$ ) $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 0.88(\mathrm{t}, 3, J=6 \mathrm{~Hz}$, $\mathrm{CH}_{3}$ ), 1.27 (br s. $8 . \mathrm{CH}_{2}$ ), $1.40-2.30\left(\mathrm{~m} .8, \mathrm{CH}_{2}\right.$ ), 2.27 (br, $1, \mathrm{OH}$ ), 3.67. :3. 70 (s. 3. OCH 3 ), 3.99-4.40 (m. 1, CHO). Anal. Caled for $\mathrm{C}_{13} \mathrm{H}_{24} \mathrm{O}_{3}$ : C. 68.38 ; H, 10.59 . Found: C, 68.20 . H. 10.53.

Methyl 1-Methyl-2-(methanesulfonoxy)cyclapentadecane1 -carboxylate ( $4 \mathrm{~b}, n=12, \mathrm{R}^{1}=\mathbf{R}^{2}=\mathbf{M e}$ ). A solution of $\mathbf{4 a}(n=12$. $\left.\mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{Me} ; 196 \mathrm{mg}, 0.66 \mathrm{mmol}\right)$ and $\mathrm{MeSO}_{2} \mathrm{Cl}(226 \mathrm{mg}, 1.97 \mathrm{mmol})$ in pyridine ( 3 mL ) was stirred for 30 min at $0-5{ }^{\circ} \mathrm{C}$ and for 3 h at room temperature. The mixture was quenched with cold water and worked up in the usual manner to give $247 \mathrm{mg}\left(100 \%\right.$ ) of $4 \mathrm{~b}\left(\mathrm{n}=12, \mathrm{R}^{1}=\mathrm{R}^{2}\right.$ $=$ Me): IR (neat) 1729 (ester $\mathrm{C}=-\mathrm{O}$ ), $1341,1171\left(\mathrm{SO}_{2}\right) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(60 \mathrm{MHz}\right.$ ) $\delta 1.20,1.24\left(\mathrm{~s}, 3, \mathrm{CH}_{3}\right.$ ), $1.30\left(\mathrm{br}, 26, \mathrm{CH}_{2}\right), 2.93,3.01(\mathrm{~s}, 3$. $\mathrm{SO}_{2} \mathrm{CH}_{3}$ ), $3.69\left(\mathrm{~s}, 3, \mathrm{OCH}_{3}\right), 4.60-5.20(\mathrm{~m}, \mathrm{l}, \mathrm{CHO})$. Anal. Caled for $\mathrm{C}_{19} \mathrm{H}_{35} \mathrm{O}_{5} \mathrm{~S}: \mathrm{C}, 60.61: \mathrm{H}, 9.64$. Found: C, 60.79; H, 9.67.
Methyl 1-methyl-2-(methanesulfonoxy)cyclododecane-1carboxylate ( $\mathbf{4 b}, n=9, \mathbf{R}^{1}=\mathbf{R}^{2}=\mathbf{M e}$ ) was prepared in the same manner as described above in $98 \%$ yield: IR (neat) $1730 \mathrm{~cm}^{-1}$ (ester): ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 1.24$ ( $\mathrm{s}, 3, \mathrm{CH}_{3}$ ), 1.38 (br s, $20, \mathrm{CH}_{2}$ ), 2.96, 3.06 $\left(\mathrm{s}, 3, \mathrm{SO}_{2} \mathrm{CH}_{3}\right), 3.70\left(\mathrm{~s}, 3 . \mathrm{OCH}_{3}\right), 4.86,5.38(\mathrm{~d}, 1, J=8 \mathrm{~Hz}, \mathrm{CHO})$. Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{31} \mathrm{O} \mathrm{O}_{5} \mathrm{~S}: \mathrm{C}, ~ 57.47$; $\mathrm{H}, 9.04$. Found: $\mathrm{C}, 57.76 ; \mathrm{H}$, 8.82 .

Methyl 2-(methanesulfonoxy)-1-pentylcycloheptane-1-car-
boxylate ( $\mathbf{4 b}, \boldsymbol{n}=\mathbf{4}, \mathbf{R}^{\mathbf{1}}=\boldsymbol{n}-\mathbf{C}_{\mathbf{5}} \mathbf{H}_{11}, \mathbf{R}^{\mathbf{2}}=\mathbf{M e}$ ) was prepared in the same manner as described above in $99 \%$ yield: IR (neat) $1735 \mathrm{~cm}^{-1}$ (ester $\mathrm{C}=0$ ); ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 0.87\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.23$ (br, 8, $\mathrm{CH}_{2}$ ), 1.40-2.36 (m, 10, $\mathrm{CH}_{2}$ ), 2.92, 2.97 ( $\mathrm{s}, 3, \mathrm{SO}_{2} \mathrm{CH}_{3}$ ), 3.66 (s, $\left.3, \mathrm{OCH}_{3}\right), 5.00,5.16(\mathrm{~d}, 1, J=6 \mathrm{~Hz}, \mathrm{CHO})$. Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{28} \mathrm{O}_{5} \mathrm{~S}$ : C, $56.23 ; \mathrm{H}, 8.81$. Found: C, $56.06 ; \mathrm{H}, 8.94$.

Ethyl 2-(methanesulfonoxy)-1-pentylcyclohexane-1-carboxylate (4b, $n=3, R^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}, \mathrm{R}^{\mathbf{2}}=\mathrm{Et}$ ) was prepared in the same manner as described above in $97 \%$ yield: IR (neat) $1730 \mathrm{~cm}^{-1}$ (ester $\mathrm{C}=\mathrm{O}$ ); ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 0.88\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right.$ ), $1.05-2.30\left(\mathrm{~m}, 16, \mathrm{CH}_{2}\right), 1.29\left(\mathrm{t}, 3, J=7 \mathrm{~Hz} . \mathrm{CH}_{3}\right.$ ), 2.98, 3.03 (s, 3 , $\left.\mathrm{SO}_{2} \mathrm{CH}_{3}\right), 4.02-4.37\left(\mathrm{~m}, 2, \mathrm{CH}_{2} \mathrm{O}\right), 5.01,5.24(\mathrm{~d}, 1, J=6 \mathrm{~Hz}, \mathrm{CHO})$. Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{28} \mathrm{O}_{5} \mathrm{~S}: \mathrm{C}, 56623 ; \mathrm{H}, 8.81$. Found: C. $56.44 ; \mathrm{H}$, 8.88 .

Methyl 1-hexyl-2-(methanesulfonoxy)cyclopentane-1-carboxylate ( $\mathbf{4 b}, \boldsymbol{n}=\mathbf{2 ,} \mathrm{R}^{1}=n-\mathrm{C}_{6} \mathrm{H}_{13}, \mathrm{R}^{2}=\mathrm{Me}$ ) was prepared in the same manner as described above in $98 \%$ yield: IR (neat) $1727 \mathrm{~cm}^{-1}$ (ester $\mathrm{C}=\mathrm{O}$ ); ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 0.86\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.26$ (br, 8, $\mathrm{CH}_{2}$ ), 1.56-2.32 (m, 8, $\mathrm{CH}_{2}$ ), 2.96, 3.02 ( $\mathrm{s}, 3, \mathrm{SO}_{2} \mathrm{CH}_{3}$ ), 3. 70 ( s , $\left.3, \mathrm{OCH}_{3}\right), 4.89,5.28(\mathrm{t}, 1, J=4 \mathrm{~Hz}, \mathrm{CHO})$. Anal. Caled for $\mathrm{C}_{14} \mathrm{H}_{26} \mathrm{O}_{5} \mathrm{~S}$ : C, 54.89; H, 8.55. Found: C, 54.87; H, 8.55 .

Methyl ( $E$ )- and ( $Z$ )-1-Methyl-2-cyclopentadecene-1-carboxylates ( $\mathbf{a}, \boldsymbol{n}=12, \mathbf{R}^{1}=\mathbf{M e}$ ). A mixture of $4 \mathbf{b}\left(n=12, \mathrm{R}^{1}=\mathrm{R}^{2}\right.$ $=\mathrm{Me} ; 220 \mathrm{mg}, 0.59 \mathrm{mmol})$ and $\mathrm{Me}_{2} \mathrm{SO}(2 \mathrm{~mL})$ was heated at $135-140$ ${ }^{\circ} \mathrm{C}$ for 3 h and worked up in the usual manner to give $162 \mathrm{mg}(99 \%)$ of la $\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right)$ : bp $73.0-75.5^{\circ} \mathrm{C}(0.0035 \mathrm{~mm}$, Kugelrohr); IR (neat) $1733 \mathrm{~cm}^{-1}$ (ester $\mathrm{C}=0$ ): ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 1.26\left(\mathrm{~s}, 3, \mathrm{CH}_{3}\right)$, 1.32 (br s, 22, $\mathrm{CH}_{2}$ ), $2.12\left(\mathrm{~m}, 2, \mathrm{CH}_{2} \mathrm{C}=\mathrm{C}\right), 3.66\left(\mathrm{~s}, 3, \mathrm{OCH}_{3}\right), 5.25-\overline{5} .75$ ( $\mathrm{m}, 2, \mathrm{HC}=\mathrm{C}$ ) Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{32} \mathrm{O}_{2}$ : C. 77.09 : $\mathrm{H}, 11.50$. Found: C, 77.02; H, 11.60 .
Methyl ( $E$ )- and ( $Z$ )-1-methyl-2-cyclododecene-1-carboxylates ( $\mathbf{a} \mathbf{a}, n=9, \mathbf{R}^{1}=\mathbf{M e}$ ) were prepared in the same manner as described above in $97 \%$ yield: bp $139.0-141.0^{\circ} \mathrm{C} / 2 \mathrm{~mm}$, Kugelrohr); IR (neat) $1730 \mathrm{~cm}^{-1}$ (ester $\mathrm{C}=0$ ); ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 1.26$ (s, $3, \mathrm{CH}_{3}$ ), 1.29 (br, 14, $\mathrm{CH}_{2}$ ), 1.61-2.30 ( $\mathrm{m}, 4, \mathrm{CH}_{2}$ ), 3.66 (s, 3, $\mathrm{OCH}_{3}$ ), 5.12-5.86 ( $\mathrm{m}, 2, \mathrm{HC}=\mathrm{C}$ ). Anal. Calcd for $\mathrm{C}_{1.5} \mathrm{H}_{26} \mathrm{O}_{2}$ : C., 75.58: H. 10.99. Found: C, 75.74; H, 11.17.
Methyl 1-pentyl-2-cycloheptene-1-carboxylate (1a, $n=4, \mathbf{R}^{1}$ $=\boldsymbol{n}-\mathbf{C}_{5} \mathbf{H}_{11}$ ) was prepared in the same manner as described above in $97 \%$ yield: bp $96.0-98.0^{\circ} \mathrm{C}$ ( 2 mm , Kugelrohr ; IR (neat) 3020,1732 (ester $\mathrm{C}=0$ ), $1650(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1},{ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 0.87(\mathrm{t}, 3 . j$ $=6 \mathrm{~Hz}, \mathrm{CH}_{3}$ ), 1.25 (br, $\left.6, \mathrm{CH}_{2}\right), 1.40-1.80\left(\mathrm{~m} .10, \mathrm{CH}_{2}\right), 1.90-2.17(\mathrm{~m}$, $\left.2, \mathrm{CH}_{2} \mathrm{C}=\mathrm{C}\right), 3.63\left(\mathrm{~s}, 3, \mathrm{OCH}_{3}\right), 5.49-5.89(\mathrm{~m} .2, \mathrm{HC}=\mathrm{C})$. Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{24} \mathrm{O}_{2}: \mathrm{C}, 74.95 ; \mathrm{H}, 10.78$. Found: C, 75.05; H, 10.84.
Ethyl 1-pentyl-2-cyclohexene-1-carboxylate (1b, $n=3, \mathbf{R}^{1}=$ $n-\mathrm{C}_{5} \mathbf{H}_{11}$ ) was prepared in the same manner as described above in $91 \%_{0}$ yield: bp $8: 3.0-85.0)^{\circ} \mathrm{C}(2 \mathrm{~mm}$, Kugelrohr); IR (neat) ; 3020,1730 lester $\mathrm{C}=0), 1651(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $\left.100 \mathrm{MH} z\right) \delta 0.88(\mathrm{t} .3, J=6 \mathrm{~Hz}$, $\mathrm{CH}_{3}$ ), $1.23\left(\mathrm{br}, 6, \mathrm{CH}_{2}\right), 1.25\left(\mathrm{t}, 3, J=7 \mathrm{~Hz},\left(\mathrm{CH}_{3}\right), 1.41-1.81(\mathrm{~m}, 5)\right.$. $\mathrm{CH}_{2}$ ), $1.95-2.28\left(\mathrm{~m}, 3, \mathrm{CH}_{2}\right), 4.17\left(\mathrm{q}, 2 . J=7 \mathrm{~Hz} . \mathrm{CH}_{v}\right.$ ) $)$, $\overline{0} .79$ (brs. ${ }^{2}, \mathrm{HC}=\mathrm{C}$. Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{24} \mathrm{O}_{2}: \mathrm{C}, 74,9 ; \mathrm{H}, 10.78$. Found: C . 75.19; H, 10.91.

Methyl 1-hexyl-2-cyclopentene-1-carboxylate (la, $n=2, \mathbf{R}^{1}$ $=n-\mathrm{C}_{6} \mathrm{H}_{13}$ ) was prepared in the same manner as described above in 98\% yield: bp 92.0-94.0 ${ }^{\circ} \mathrm{C}(3 \mathrm{~mm}$, Kugelrohr): IR (neat) 3040, 173.2 (ester $\mathrm{C}=0$ ), $1619(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-3} ;{ }^{1} \mathrm{H}$ NMR ( 1010 MHz ) $\delta 0.88(\mathrm{t}, 3 . \mathrm{J}$ $\left.=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.28\left(\mathrm{br}, 8 . \mathrm{CH}_{2}\right), 1.50-1.97\left(\mathrm{~m} .3, \mathrm{CH}_{2}\right), 2.24-2.56(\mathrm{~m}$. 3, $\mathrm{CH}_{2}$ ), 3.72 ( $\mathrm{s}, 3, \mathrm{OCH}_{3}$ ), $5.68-5.94$ (m. 2. $\mathrm{H}(=\mathrm{C})$. Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{2}: \mathrm{C}, 74,24 ; \mathrm{H}, 10.54$. Found: C, 74.20 : H. 10.60.
( $E$ )- and (Z)-1-Methyl-2-cyclopentadecene-1-carboxylic Acids ( $\mathbf{1 c}, \boldsymbol{n}=12, \mathbf{R}^{1}=\mathbf{M e}$ ). A solution of $1 \mathbf{a}\left(n=12 . \mathrm{R}^{1}=\mathrm{Me} ; 240.5\right.$ $\mathrm{mg}, 0.86 \mathrm{mmol}$ ) and KOH ( $290 \mathrm{mg}, 5.17 \mathrm{mmol}$ ) in MeOH ( 7 mL ) and water ( 0.6 mL ) was stirred for 5 h at $60-65^{\circ} \mathrm{C}$. Atter the solvent was removed, the mixture was washed with benzene and the aqueous laver was acidified with cold aqueous $10 \% \mathrm{HCl}$ and worked up in the usual manner to give $218 \mathrm{mg}(95 \%)$ of $1 \mathrm{c}(\mathrm{n}=12 . \mathrm{K}=$ Me) as a solid: mp 69.5-71.5 ${ }^{\circ} \mathrm{C}$; IR ( Nujol$) 3400-2600(\mathrm{COOH}), 15(0)(\mathrm{COOH}) \mathrm{cm}^{-1},{ }^{\prime} \mathrm{H}$ NMR $(100 \mathrm{MHz}) \delta 1.31\left(\mathrm{~s}, 3, \mathrm{CH}_{3}\right), 1.33$ (brs. 22. $\left(\mathrm{CH}_{2}\right) \cdot 2.00-2.32(\mathrm{~m}$. $\left.2, \mathrm{CH}_{2} \mathrm{C}=\mathrm{C}\right), 5.40-5.72(\mathrm{~m}, 2, \mathrm{HC}=\mathrm{C}), 11.40(\mathrm{br}, 1, \mathrm{COOH})$. Anal. (aled for $\mathrm{C}_{17} \mathrm{H}_{35} \mathrm{O}_{2}: \mathrm{C}, 76.64 ; \mathrm{H}, 11: 35$. Found: ( $\mathrm{C}, 76.75: \mathrm{H}, 11.57$.
( $E$ )- and ( $Z$ )-1-Methyl-2-cyclododecene-1-carboxylic acids ( $\mathbf{1 c}, \boldsymbol{n}=9, \mathrm{R}^{\mathbf{1}}=\mathrm{Me}$ ) were prepared in the same manner as described above in $86 \%$ yield: IR (neat) $3600-2500(\mathrm{COOH}) .1697(\mathrm{COOH}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 1.26 .1 .36\left(\mathrm{~s}, 3 . \mathrm{CH}_{3}\right), 1.31\left(\mathrm{brs}, 16 . \mathrm{CH}_{2}\right)$. $1.90-2.29\left(\mathrm{~m}, 2, \mathrm{CH}_{2} \mathrm{C}=\mathrm{C}\right), 5.10-5.84(\mathrm{~m}, 2 . \mathrm{HC}=(\mathrm{O}) .11 .76(\mathrm{br} .1$. $(\mathrm{OOH})$. Anal. Caled for $\mathrm{C}_{14} \mathrm{H}_{24} \mathrm{O}_{2}: \mathrm{C} .74 .95 ; \mathrm{H}, 10.78$. Found: (., 74.88: H. 10.83 .

1-Pentyl-2-cycloheptene-1-carboxylic acid (1c, $n=4, \mathrm{R}^{1}=$ $n-\mathrm{C}_{5} \mathbf{H}_{11}$ ) was prepared in the same manner as described above in $88 \%$ vield: IR (neat) $3400-2400(\mathrm{COOH}), 1697(\mathrm{COOH}) \mathrm{cm}^{-1}:{ }^{1} \mathrm{H}$ NMR
$(100 \mathrm{MHz}) \delta 0.87\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.28\left(\mathrm{br}, 6, \mathrm{CH}_{2}\right), 1.48-1.92(\mathrm{~m}$, $10, \mathrm{CH}_{2}$ ), 2.00-2.22 (m, 2, $\left.\mathrm{CH}_{2} \mathrm{C}=\mathrm{C}\right), 5.50-6.00(\mathrm{~m}, 2, \mathrm{HC}=\mathrm{C}), 10.18$ (br, 1, COOH ). Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{2}$ : $\mathrm{C}, 74.24 ; \mathrm{H}, 10.54$. Found: C, $74.49 ; \mathrm{H}, 10.70$.
1-Pentyl-2-cyclohexene-1-carboxylic acid (1c, $n=3, R^{1}=$ $\boldsymbol{n}-\mathrm{C}_{5} \mathrm{H}_{\mathbf{1 1}}$ ) was prepared in the same manner as described above in $93 \%$ yield: IR (neat) $3400-2400(\mathrm{COOH}), 1695(\mathrm{COOH}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $(100 \mathrm{MHz}) \delta 0.91\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.30\left(\mathrm{br}, 6, \mathrm{CH}_{2}\right), 1.46-1.79(\mathrm{~m}$, $5, \mathrm{CH}_{2}$ ), 1.88-2.30 (m, 3, $\mathrm{CH}_{2}$ ), $5.76-6.04(\mathrm{~m}, 2, \mathrm{HC}=\mathrm{C}), 10.70(\mathrm{br}, 1$, $\mathrm{COOH})$. Anal. Caled for $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{O}_{2}: \mathrm{C}, 73.43 ; \mathrm{H}, 10.27$. Found: C, 73.43 ; H, 10.38 .

1-Hexyl-2-cyclopentene-1-carboxylic acid (1c, $n=2, R^{1}=$ $n-\mathbf{C}_{6} \mathbf{H}_{13}$ ) was prepared in the same manner as described above in $83 \%$ yield: IR (neat) $3400-2400(\mathrm{COOH}), 1698(\mathrm{COOH}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $(100 \mathrm{MHz}) \delta 0.90\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.28\left(\mathrm{br}, 8, \mathrm{CH}_{2}\right), 1.48-1.99(\mathrm{~m}$, $\left.3, \mathrm{CH}_{2}\right), 2.44\left(\mathrm{~m}, 3, \mathrm{CH}_{2}\right), 5.76-5.96(\mathrm{~m}, 2, \mathrm{HC}=\mathrm{C}), 9.95(\mathrm{br}, 1, \mathrm{COOH})$. Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{O}_{2}$ : C, $73.43 ; \mathrm{H}, 10.27$. Found: C, $73.40 ; \mathrm{H}$, 10.46 .

8,8a-Epoxy-2,2-(ethylenedioxy)-4a-decalylcarboxylic Acid (6). To a solution of $8 \mathbf{b}$ ( $105 \mathrm{mg}, 0.44 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 5 mL ) was added $85 \%$-chloroperbenzoic acid ( $120 \mathrm{mg}, 0.59 \mathrm{mmol}$ ), and the mixture was stirred for 20 h at $5^{\circ} \mathrm{C}$. Removal of the solvent and subsequent chromatography ( $\mathrm{SiO}_{2}, 1: 1$ hexane- AcOEt ) gave 99.6 mg ( $89 \%$ ) of 6 as an oil: IR (neat) $3600-2400(\mathrm{COOH}), 1710(\mathrm{COOH}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 1.19-3.04\left(\mathrm{~m}, 13, \mathrm{CH}_{2}, \mathrm{CHO}\right), 3.97-4.16(\mathrm{~m}, 4$, $\mathrm{CH}_{2} \mathrm{O}$ ), 6.16 (br s, 1, COOH ). Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{5}: \mathrm{C}, 61.41$; H , 7.14. Found: C, 61.51; H, 7.20.

2,2-(Ethylenedioxy)-1,2,3,4,4a,5,6,7-octahydronaphthalene-4a-carboxylic Acid (8b). To a solution of $8 \mathbf{a}^{10}(416 \mathrm{mg}, 1.56 \mathrm{mmol})$ in MeOH ( 15 mL ) was added a solution of KOH ( $530 \mathrm{mg}, 9.46 \mathrm{mmol}$ ) in water ( 1 mL ). The mixture was stirred for 24 h at $55-60^{\circ} \mathrm{C}$ and worked up in the usual manner to give 292 mg ( $79 \%$ ) of 8 b as a solid: $\mathrm{mp} 71.5-73.5^{\circ} \mathrm{C}$; IR (Nujol) $3400-2500(\mathrm{COOH}), 1680(\mathrm{COOH}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 1.30-3.05\left(\mathrm{~m}, 12, \mathrm{CH}_{2}\right), 3.96\left(\mathrm{~s}, 4, \mathrm{CH}_{2} \mathrm{O}\right), 5.69$ (br s, 1, $\mathrm{HC}=\mathrm{C}$ ), 9.56 (br s. $1, \mathrm{COOH}$ ). Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{4}: \mathrm{C}$, $65.53 ;$ H, 7.61. Found: C, 65.80; H, 7.88.

Electrolysis Apparatus. An undivided cell was equipped with a platinum anode $\left(3 \mathrm{~cm}^{2}\right)$, a platinum cathode ( $3 \mathrm{~cm}^{2}$ ), and/or a SUS-27 stainless steel cathode ( $11 \mathrm{~cm}^{2}$ ), a gas lead pipe, and a thermometer. Regulated dc power was supplied by a Metronix 543B instrument.

General Procedure for Electrochemical Synthesis of 2a from le in $\mathrm{AcOH}-\mathrm{AcOEt}-\boldsymbol{t}-\mathrm{BuOH}$. A stirred solution of lc ( $n=12, \mathrm{R}^{1}$ $=\mathrm{Me} ; 26 \mathrm{mg}, 0.098 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(436 \mathrm{mg}, 4.3 \mathrm{mmol})$ in $\mathrm{AcOH}(1.5$ $\mathrm{mL}), \mathrm{AcOEt}(3.5 \mathrm{~mL})$, and $t-\mathrm{BuOH}(0.19 \mathrm{~mL})$ was electrolyzed in a heaker fitted with two platinum electrodes at a constant applied voltage of 30 V , current density $36-54 \mathrm{~mA} / \mathrm{cm}^{2}$, for $3 \mathrm{~h}(\sim 153 \mathrm{~F} / \mathrm{mol}$ ) at $19-22^{\circ} \mathrm{C}$. After electrolysis, the mixture was concentrated and taken up in ether. The extracts were washed with brine, dried ( $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ), and shaken over solid $\mathrm{K}_{2} \mathrm{CO}_{3}$. Removal of the solvent and subsequent chromatography ( $\mathrm{SiO}_{2}, 5: 1$ hexane-AcOEt) gave 21.8 mg $(79 \%)$ of $2 \mathrm{a}\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right)$ as an oil: bp $110.0-111.0^{\circ} \mathrm{C}(0.03 \mathrm{~mm}$, Kugelrohr): IR (neat) 1735 (ester $\mathrm{C}==0$ ), $1670(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR $(100) \mathrm{MHz}) \delta 1.32$ (br, $24, \mathrm{CH}_{2}$ ), 1.70 (s, $3, \mathrm{CH}_{3}$ ), 2.01 (s, $3, \mathrm{COCH}_{3}$ ), $2.00-2.18\left(\mathrm{~m}, 2, \mathrm{CH}_{2}\right), 5.12(\mathrm{~d}, 1, J=10 \mathrm{~Hz}, \mathrm{HC}=\mathrm{C}), 5.38-5.60(\mathrm{~m}$, 1, (HO). Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{32} \mathrm{O}_{2}: \mathrm{C}, 77.09$; $\mathrm{H}, 11.50$. Found: $\mathrm{C}, 77.25$; H. 11.41.
( $E$ )- and ( $Z$ )-;-Methyl-2-cyclododecen-1-ol Acetates (2a, $n$ $=9, \mathrm{R}^{1}=\mathbf{M e}$ ): $\mathrm{bp} 45.0-47.0^{\circ} \mathrm{C}(0.0015 \mathrm{~mm}$, Kugelrohr); IR (neat) 1735 (ester $\mathrm{C}=\mathrm{O}), 1660(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $(100 \mathrm{MHz}) \delta 1.28$ (br, 16, $\mathrm{CH}_{2}$ ), 1.53. 1.73 ( $\mathrm{s}, 3, \mathrm{CH}_{3}$ ), $1.90-2.10\left(\mathrm{~m}, 2, \mathrm{CH}_{2}\right), 2.01(\mathrm{~s}, 3$, $\left(\mathrm{OCH}_{3}\right), 5.26(\mathrm{~d}, 1, J=10 \mathrm{~Hz}, \mathrm{HC}=\mathrm{C}), 5.34-5.65(\mathrm{~m}, 1, \mathrm{CHO})$. Anal. Caled for $\mathrm{C}_{15} \mathrm{H}_{26} \mathrm{O}_{2}: \mathrm{C}, 75.58 ; \mathrm{H}, 10.99$. Found: C, $75.53 ; \mathrm{H}, 11.07$.

3-Pentyl-2-cyclohepten-1-ol Acetate (2a, $n=4, \mathbf{R}^{1}=\boldsymbol{n}-\mathbf{C}_{5} \mathbf{H}_{11}$ ): bp $57.0-59.0^{\circ} \mathrm{C}(0.002 \mathrm{~mm}$. Kugelrohr); IR (neat) 1730 (ester $\mathrm{C}=0$ ), 1370. 1241, $1025 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 0.90(\mathrm{t}, 3, J=6 \mathrm{~Hz}$, $\left(\mathrm{H}_{3}\right), 1.20-2.24\left(\mathrm{~m}, 10, \mathrm{CH}_{2}\right), 1.29\left(\mathrm{br}, 6, \mathrm{CH}_{2}\right), 2.03\left(\mathrm{~s}, 3, \mathrm{COCH}_{3}\right)$, 3. 41 (br s, $2 . \mathrm{HC}=\mathrm{C}, \mathrm{CHO}$; ${ }^{13} \mathrm{C}$ NMR $\delta 14.0$ (q), 21.4 (q, acetyl $\mathrm{CH}_{3}$ ), $29.6(\mathrm{t}), 26.1(\mathrm{t}), 27.2(\mathrm{t}), 27.3(\mathrm{t}), 31.5(\mathrm{t}), 32.5(\mathrm{t}), 32.9(\mathrm{t}), 40.1(\mathrm{t}, \mathrm{C}-7)$, $74.0(\mathrm{~d}, \mathrm{C}-1), 127.0(\mathrm{~d}, \mathrm{C}-2), 144.0(\mathrm{~s}, \mathrm{C}-3), 170.4$ (s, acetyl $\mathrm{C}=\mathrm{O}$ ). Anal. (aled for $\mathrm{C}_{14} \mathrm{H}_{24} \mathrm{O}_{2}$ : C, $74.95 ; \mathrm{H}, 10.78$. Found: C, $75.11 ; \mathrm{H}, 10.76$.

3-Pentyl-2-cyclohexen-1-ol Acetate (2a, $n=3, \mathrm{R}^{1}=\boldsymbol{n}-\mathrm{C}_{5} \mathbf{H}_{11}$ ): bp $51.0-53.0^{\circ} \mathrm{C}(0.011 \mathrm{~mm}$, Kugelrohr); IR (neat) 1730 (ester $\mathrm{C}=\mathrm{O}$ ), $1664(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $(100 \mathrm{MHz}) \delta 0.89\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right)$, 1.29 (br, $6, \mathrm{CH}_{2}$ ), 1.71 (br, $4, \mathrm{CH}_{2}$ ), 1.91-2.40 (m, 4, $\mathrm{CH}_{2}$ ), 2.03 (s, 3 , $\mathrm{COCH}_{3}$ ), 5.27 (br, $1, \mathrm{HC}==\mathrm{C}$ ), 5.44 ( $\mathrm{br} \mathrm{s}, 1, \mathrm{CHO}$ ); ${ }^{13} \mathrm{C}$ NMR $\delta 14.0(\mathrm{q})$, $19.2(\mathrm{t}), 21.5\left(\mathrm{q}\right.$, acetyl $\left.\mathrm{CH}_{3}\right), 22.6(\mathrm{t}), 27.1(\mathrm{t}), 28.4(\mathrm{t}, 2 \mathrm{C}), 31.6(\mathrm{t}, \mathrm{C}-4)$, 37.7 (t, C-6), $68.9(\mathrm{~d}, \mathrm{C}-1), 119.2$ (d, C-2), 144.9 (s, C-3), 170.8 (s, acetyl $\mathrm{C}=\mathrm{O}$. Anal. Caled for $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{2}: \mathrm{C}, 74.24 ; \mathrm{H}, 10.54$. Found: $\mathrm{C}, 74.39$; H, 10.62 .

3-Hexyl-2-cyclopenten-1-ol Acetate (2a, $n=2, \mathrm{R}^{1}=\boldsymbol{n}-\mathrm{C}_{6} \mathrm{H}_{13}$ ): bp $49.0-51.0^{\circ} \mathrm{C}(0.006 \mathrm{~mm}$. Kugelrohr); IR (neat) 1730 (ester $\mathrm{C}=\mathrm{O}$ ),
$1682(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 100 MHz$) \delta 0.90\left(\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right)$, $1.31\left(\mathrm{br}, 8, \mathrm{CH}_{2}\right), 1.50-2.55\left(\mathrm{~m}, 6, \mathrm{CH}_{2}\right), 1.92\left(\mathrm{~s}, 3, \mathrm{COCH}_{3}\right), 5.41$ (complex s, 1, $\mathrm{HC}=\mathrm{C}$ ), 5.52 (broad m, 1, CHO ). Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{2}$ : C, $74.24 ; \mathrm{H}, 10.54$. Found: C, $74.35 ; \mathrm{H}, 10.45$.
General Procedure for Electrosynthesis of 2 c and/or 5 from lc in MeOH. A stirred solution of $\mathbf{1 c}\left(n=12, \mathrm{R}^{1}=\mathrm{Me} 27.8 \mathrm{mg}, 0.104\right.$ mmol ) in $\mathrm{MeOH}(15 \mathrm{~mL})$ containing $\mathrm{Et}_{3} \mathrm{~N}$ ( $210 \mathrm{mg}, 2.08 \mathrm{mmol}$ ) was electrolyzed in a beaker fitted with a platinum anode and a SUS-27 stainless steel cathode at a constant applied voltage of 20 V , current density $41-44 \mathrm{~mA} / \mathrm{cm}^{2}$, for $3 \mathrm{~h}(\sim 140 \mathrm{~F} / \mathrm{mol})$ at $20-26^{\circ} \mathrm{C}$. The electrolysis solution was concentrated, and the residue was taken up in ether-benzene. The organic layer was washed with brine and dried ( $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ). Removal of the solvent and subsequent chromatography ( $\mathrm{SiO}_{2}, 7: 1$ hexane-AcOEt) gave $21.9 \mathrm{mg}(83 \%)$ of a mixture of 2 c and $5\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right)$ [the ratio of $\mathbf{2 c} / 5$ was estimated to be $1: 2$ based on ${ }^{1} \mathrm{H}$ NMR signals at $\delta 3.16\left(\mathrm{~s}, \mathrm{OCH}_{3}\right)$ and $\left.3.22\left(\mathrm{~s}, \mathrm{OCH}_{3}\right)\right]$ : bp $70.5-73.0^{\circ} \mathrm{C}(0.003 \mathrm{~mm}$, Kugelrohr); IR (neat) $1660(\mathrm{C}=\mathrm{C}), 1458$, $1371,1088,976 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 1.20,1.68\left(\mathrm{~s}, 3, \mathrm{CH}_{3}\right), 1.30$ ( $\mathrm{br}, 22, \mathrm{CH}_{2}$ ), $2.13\left(\mathrm{~m}, 2, \mathrm{CH}_{2}\right), 3.16,3.22\left(\mathrm{~s}, 3, \mathrm{OCH}_{3}\right), 3.94(\mathrm{~m}, \mathrm{CHO})$, 4.99-5.67 (m, $\mathrm{HC}=\mathrm{C}$ ). Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{32} \mathrm{O}: \mathrm{C}, 80.89$; H, 12.78 . Found: C, 80.97; H, 12.91.
$(E)$ - and (Z)-3-Methoxy-1-methyl-2-cyclododecenes (2c, $n=$ 9, $\mathbf{R}^{1}=\mathbf{M e}$ ) and/or $(E)$ - and ( $Z$ )-1-methoxy-1-methyl-2-cyclododecenes $\left(5, \boldsymbol{n}=9, \mathbf{R}^{1}=\mathbf{M e}\right)(2 \mathrm{c} / 5=1: 2)$ : bp $75.0-77.5^{\circ} \mathrm{C}(0.005$ mm , Kugelrohr); IR (neat) $1660(\mathrm{C}=\mathrm{C}), 1460,1443,1371,1080,982$ $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $(100 \mathrm{MHz}) \delta 1.21,1.71\left(\mathrm{~s}, 3, \mathrm{CH}_{3}\right), 1.29\left(\mathrm{br}, 16, \mathrm{CH}_{2}\right)$, $2.10\left(\mathrm{~m}, 2, \mathrm{CH}_{2}\right), 3.19,3.26\left(\mathrm{~s}, 3, \mathrm{OCH}_{3}\right), 3.79-4.04(\mathrm{~m}, \mathrm{CHO}), 5.12-5.48$ (m, $\mathrm{HC}=\mathrm{C}$ ). Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{26} \mathrm{O}: \mathrm{C}, 79.94 ; \mathrm{H}, 12.46$. Found: C, 79.91; H, 12.27

1-Methoxy-3-pentyl-2-cyclohexene (2c, $n=3, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}$ ): bp $73.5-75.0^{\circ} \mathrm{C}(0.006 \mathrm{~mm}$, Kugelrohr); IR (neat) $1666(\mathrm{C}=\mathrm{C}), 1468$, $1452,1378,1354,1194,1095,934,912 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 0.89$ ( $\mathrm{t}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}$ ), $1.14-2.03\left(\mathrm{~m}, 14, \mathrm{CH}_{2}\right), 3.37\left(\mathrm{~s}, 3, \mathrm{OCH}_{3}\right), 3.72$ (br, 1, CHO), 5.50 (br, 1, $\mathrm{HC}=\mathrm{C}$ ). Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{22} \mathrm{O}: \mathrm{C}, 79.06$; H, 12.16. Found: C, 79.11; H, 12.37.

1-Methoxy-1-pentyl-2-cyclohexene ( $5, n=3, R^{1}=\boldsymbol{n}-\mathbf{C}_{5} \mathbf{H}_{11}$ ): bp $72.0-74.5^{\circ} \mathrm{C}$ ( 0.006 mm , Kugelrohr); IR (neat) 3010,1648 ( $\mathrm{C}=\mathrm{C}$ ), $1468,1454,1074,908 \mathrm{~cm}^{-1}{ }^{1}{ }^{1} \mathrm{H}$ NMR $(100 \mathrm{MHz}) \delta 0.98(\mathrm{t}, 3, J=6 \mathrm{~Hz}$, $\left.\mathrm{CH}_{3}\right), 1.10-2.16\left(\mathrm{~m}, 14, \mathrm{CH}_{2}\right), 3.17,3.18\left(\mathrm{~s}, 3, \mathrm{OCH}_{3}\right), 5.57(\mathrm{~d}, 1, J=$ $10 \mathrm{~Hz}, \mathrm{HC}=\mathrm{C}$ ), 5.88 (dt, $1, J=10$ and $4 \mathrm{~Hz}, \mathrm{HC}=\mathrm{C}$ ). Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{22} \mathrm{O}: \mathrm{C}, 79.06 ; \mathrm{H}, 12.16$. Found: C, $79.26 ; \mathrm{H}, 12.28$.

Electrolysis of 6 in MeOH . A solution of $6(77.6 \mathrm{mg}, 0.306 \mathrm{mmol})$ in $\mathrm{MeOH}(12 \mathrm{~mL})$ containing $\mathrm{Et}_{3} \mathrm{~N}$ ( $210 \mathrm{mg}, 2.08 \mathrm{mmol}$ ) was electrolyzed with two platinum electrodes at a constant applied voltage of 20 V , current density $23-32 \mathrm{~mA} / \mathrm{cm}^{2}$, for $7 \mathrm{~h}(\sim 120 \mathrm{~F} / \mathrm{mol})$ at $44-49$ ${ }^{\circ} \mathrm{C}$. The mixture was concentrated, and the residue was worked up in the usual manner to give $53.5 \mathrm{mg}(84.2 \%)$ of 7 as an oil after chromatography ( $\mathrm{SiO}_{2}, 10: 1$ hexane-AcOEt): bp $73.5-76.0^{\circ} \mathrm{C}(0.007 \mathrm{~mm}$, Kugelrohr); IR (neat) $1664(\mathrm{C}=\mathrm{O}), 1635(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 60 MHz ) $\delta 1.20-2.60\left(\mathrm{~m}, 12, \mathrm{CH}_{2}\right), 3.97$ (complex s, $4, \mathrm{CH}_{2} \mathrm{O}$ ); ${ }^{13} \mathrm{C}$ NMR $\delta 22.2(\mathrm{t}), 30.7(\mathrm{t}, 2 \mathrm{C}), 31.1(\mathrm{t}), 32.3(\mathrm{t}), 37.3(\mathrm{t}), 64.4(\mathrm{t}, 2 \mathrm{C}), 107.9(\mathrm{~s}$, C-2), 130.0 ( $\mathrm{s}, \mathrm{C}-8 \mathrm{a}$ ), 155.7 ( $\mathrm{s}, \mathrm{C}-4 \mathrm{a}$ ), 198.4 ( $\mathrm{s}, \mathrm{C}-8$ ). Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{16} \mathrm{O}_{3}$ : C, $69.21 ; \mathrm{H}, 7.74$. Found: C, $69.34 ; \mathrm{H}, 7.76$

Electrolysis of $8 \mathbf{b}$ in $\mathbf{M e O H}$. A solution of $8 \mathbf{b}(102.5 \mathrm{mg}, 0.43$ $\mathrm{mmol})$ in $\mathrm{MeOH}(12 \mathrm{~mL})$ and $\mathrm{Et}_{3} \mathrm{~N}(291 \mathrm{mg}, 2.88 \mathrm{mmol})$ was electrolyzed using two platinum electrodes at a constant applied voltage of 20 V , current density $23-33 \mathrm{~mA} / \mathrm{cm}^{2}$, for $5 \mathrm{~h}(\sim 82 \mathrm{~F} / \mathrm{mol})$. The usual workup and subsequent chromatography ( $\mathrm{SiO}_{2}, 10: 1$ hexane- AcOEt ) gave $27.2 \mathrm{mg}(28.6 \%)$ of 9 and $37.8 \mathrm{mg}(39.7 \%)$ of 10 as a oil product. Physical constants together with elemental analyses of 9 and 10 are as follows.

2,2-(Ethylenedioxy)-4a-methoxy-1,2,3,4,4a,5,6,7-octahydro-
naphthalene (9): bp $41.0-42.5^{\circ} \mathrm{C}(0.01 \mathrm{~mm}$, Kugelrohr); IR (neat) 3050, 1376, 1311, 1265, 1202, 1167, 1130, 1088, 1068, 1000, 974, 965, $948,869,811 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 60 MHz ) $\delta 1.23-2.75\left(\mathrm{~m}, 12, \mathrm{CH}_{2}\right.$ ), 3.17 (s, 3, $\mathrm{OCH}_{3}$ ), $3.93\left(\mathrm{~s}, 4, \mathrm{CH}_{2} \mathrm{O}\right), 5.69(\mathrm{br} \mathrm{s}, 1, \mathrm{HC}=\mathrm{C}) ;{ }^{13} \mathrm{C}$ NMR $\delta 19.9$ ( t ), 25.3 ( t ), $30.5(\mathrm{t}), 31.6(\mathrm{t}), 33.7(\mathrm{t}), 41.2(\mathrm{t}, \mathrm{C}-1), 49.5\left(\mathrm{q}, \mathrm{OCH}_{3}\right), 64.3$ (t, 2C), 73.0 ( $\mathrm{s}, \mathrm{C}-4 \mathrm{a}$ ), 109.2 ( $\mathrm{s}, \mathrm{C}-2$ ), 127.4 (d, C-8), 135.4 ( $\mathrm{s}, \mathrm{C}-8 \mathrm{a})$. Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{33}: \mathrm{C}, 69.61 ; \mathrm{H}, 8.99$. Found: $\mathrm{C}, 69.63 ; \mathrm{H}$, 8.89

2,2-(Ethylenedioxy)-8-methoxy-1,2,3,4,5,6,7,8-octahydronaphthalene ( 10 ): bp $41.0-43.5^{\circ} \mathrm{C}(0.006 \mathrm{~mm}$, Kugelrohr); IR (neat) 1365, 1329, 1251, 1215, 1197, 1141, 1130, 1098, 1080, 1017, 947, 900, $842 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{NMR}(60 \mathrm{MHz}) \delta 1.27-2.73\left(\mathrm{~m}, 12, \mathrm{CH}_{2}\right), 3.32(\mathrm{~s}, 3$, $\left.\mathrm{OCH}_{3}\right), 3.45(\mathrm{~m}, 1, \mathrm{CHO}), 3.97\left(\mathrm{~s}, 4, \mathrm{CH}_{2} \mathrm{O}\right) ;{ }^{13} \mathrm{C}$ NMR $\delta 18.6(\mathrm{t}), 26.6$ (t), 29.7 (t), $29.9(\mathrm{t}), 31.0(\mathrm{t}), 37.5(\mathrm{t}, \mathrm{C}-1), 59.4\left(\mathrm{q}, \mathrm{OCH}_{3}\right), 64.3(\mathrm{t}, 2 \mathrm{C})$ 77.8 (d, C-8), 108.6 (s, C-2), 126.5 (s, C-4a), 132.4 (s, C-8a). Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{3}: \mathrm{C}, 69.61 ; \mathrm{H}, 8.99$. Found: C, $69.69 ; \mathrm{H}, 8.90$.
( $E$ )- and ( $Z$ )-3-Methyl-2-cyclopentadecen-1-ols ( $2 \mathrm{~b}, \boldsymbol{n}=12$, $\mathbf{R}^{1}=\mathbf{M e}$ ). A solution of $\mathbf{2 a}\left(n=12, \mathrm{R}^{1}=\mathrm{Me} ; 20 \mathrm{mg}, 0.071 \mathrm{mmol}\right)$ and KOH ( $50 \mathrm{mg}, 0.89 \mathrm{mmol}$ ) in $25 \%$ aqueous $\mathrm{MeOH}(2 \mathrm{~mL})$ was stirred
for 8 h at $5-10^{\circ} \mathrm{C}$. The mixture was worked up in the usual manner to give $16 \mathrm{mg}(93 \%)$ of $\mathbf{2 b}\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right)$ as an oil: bp $51.5-53.5^{\circ} \mathrm{C}$ ( 0.005 mm , Kugelrohr); IR (neat) $3320(\mathrm{OH}), 1668(\mathrm{C}=\mathrm{C}), 1460,1380$, $1015,720 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 1.32$ (br s, $23, \mathrm{CH}_{2}, \mathrm{OH}$ ), 1.68 (s, 3, $\mathrm{CH}_{3}$ ), 2.04-2.15 (m, 2, $\mathrm{CH}_{2}$ ), 4.28-4.54 (m, 1, CHO), 5.18 (d, 1, $J=10 \mathrm{~Hz}, \mathrm{HC}=\mathrm{C}$ ). Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{30} \mathrm{O}: \mathrm{C}, 80.61 ; \mathrm{H}, 12.68$. Found: C, 80.54 ; H, 12.55.
$(E)$ - and (Z)-3-Methyl-2-cyclopentadecen-1-ones (11). ${ }^{11} \mathrm{To}$ a stirred suspension of pyridinium chlorochromate ( $30 \mathrm{mg}, 0.11 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ was added dropwise a solution of $2 \mathrm{~b}\left(n=12, \mathrm{R}^{1}=\right.$ $\mathrm{Me} ; 15 \mathrm{mg}, 0.063 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $0-5^{\circ} \mathrm{C}$. The mixture was stirred for 30 min at $5^{\circ} \mathrm{C}$ and for 2 h at room temperature and worked up in the usual manner to give $9.7 \mathrm{mg}(65 \%)$ of 11 as an oil: bp $75.0-78.0^{\circ} \mathrm{C}$
 $1685(\mathrm{C}=0), 1617(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1}{ }^{1} \mathrm{H}$ NMR ( 100 MHz ) $\delta 1.30(\mathrm{br}, 22$, $\left.\mathrm{CH}_{2}\right), 2.14\left(\mathrm{~s}, 3, \mathrm{CH}_{3}\right), 2.28,2.41\left(\mathrm{~d}, 2, J=5 \mathrm{~Hz}, \mathrm{COCH}_{2}\right), 6.17(\mathrm{br} \mathrm{s}$, $1, \mathrm{HC}=\mathrm{C}$ ).
d $I$-Muscone. ${ }^{12}$ A mixture of $11(7 \mathrm{mg}, 0.03 \mathrm{mmol})$ and $10 \%$ palladium on charcoal in EtOH ( 1.5 mL ) was treated with excess $\mathrm{H}_{2}$ ( 2 mL ). The mixture was filtered, and the filtrate was concentrated to give $6.8 \mathrm{mg}(97 \%)$ of dl -muscone: ${ }^{13}$ bp $77.5-79.0^{\circ} \mathrm{C}(0.005 \mathrm{~mm}$, Kugelrohr) [lit. ${ }^{11 \mathrm{a}} 100-105^{\circ} \mathrm{C}(0.01 \mathrm{~mm})$ ]; IR (neat) $1715 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR $(60 \mathrm{MHz}) \delta 0.92\left(\mathrm{~d}, 3, J=6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.29\left(\mathrm{br}, 23, \mathrm{CH}_{2}, \mathrm{CH}\right), 2.00-2.52$ ( $\mathrm{m}, 4, \mathrm{CH}_{2}$ ).

Registry No.-( $E$ )- 1a ( $n=12, \mathrm{R}^{\mathrm{l}}=\mathrm{Me}$ ), 69832-58-2; $(Z)$-1a ( $n$ $\left.=12, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-59-3 ;(E)-\mathrm{la}\left(n=9, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-60-6$; $(Z)-1 \mathbf{a}\left(n=9, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-61-7 ; \mathbf{l} \mathbf{a}\left(n=4, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}\right)$, 69832-62-8; la $\left(n=2, \mathrm{R}^{1}=n-\mathrm{C}_{6} \mathrm{H}_{13}\right), 69832-63-9 ; \mathbf{1 b}\left(n=3, \mathrm{R}^{1}=\right.$ $\left.n-\mathrm{C}_{5} \mathrm{H}_{11}\right), 69832-64-0 ;(E)-1 \mathbf{c}\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-65-1 ;(Z)-1 \mathbf{c}$ ( $n=12, \mathrm{R}^{1}=\mathrm{Me}$ ), 69832-66-2; $(E)-\mathrm{lc}\left(n=9, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-67-3$; $(Z)-1 \mathbf{c}\left(n=9, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-68-4 ; 1 \mathbf{c}\left(n=4, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}\right)$, 69832-69-5; 1c $\left(n=3, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}\right), 69855-40-9 ; \mathrm{cc}\left(n=2, \mathrm{R}^{1}=\right.$ $n-\mathrm{C}_{6} \mathrm{H}_{13}$ ), 69832-70-8; (E)-2a ( $n=12, \mathrm{R}^{1}=\mathrm{Me}$ ), 69832-71-9; (Z)-2a ( $n=12, \mathrm{R}^{1}=\mathrm{Me}$ ), 69832-72-0; $(E)$-2a $\left(n=9, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-73-1$; ( $Z$ )-2a $\left(n=9, \mathrm{R}^{1}=\mathrm{Me}\right)$, 69832-74-2; 2a $\left(n=4, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}\right)$, 69832-75-3; $\mathbf{2 a},\left(n=3, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}\right), 69832-76-4 ; \mathbf{2 a}\left(n=2, \mathrm{R}^{1}=\right.$ $\left.n-\mathrm{C}_{6} \mathrm{H}_{13}\right), 69832-77-5 ;(E)-2 \mathbf{b}\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-78-6 ;(Z)-2 \mathrm{~b}$ $\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-79-7$; $(E)$-2c $\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-80-0$; $(Z)-2 \mathrm{c},\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-81-1 ;(E)-2 \mathrm{c}\left(n=9, \mathrm{R}^{1}=\mathrm{Me}\right)$, 69832-82-2; $(Z)$ 2c, $\left(n=9, \mathrm{R}^{1}=\mathrm{Me}\right), 69832-83-3 ; \mathbf{2 c}\left(n=3, \mathrm{R}^{1}=\right.$ $\left.n-\mathrm{C}_{5} \mathrm{H}_{11}\right), 69832-84-4 ; 3 \mathbf{a}\left(n=12, \mathrm{R}^{2}=\mathrm{Me}\right), 52794-21-5 ; 3 \mathbf{a}(n=9$, $\left.\mathrm{R}^{2}=\mathrm{Me}\right), 62939-87-1 ; 3 \mathrm{a},\left(n=4, \mathrm{R}^{2}=\mathrm{Me}\right), 52784-32-4 ; 3 \mathrm{a}\left(n=2, \mathrm{R}^{2}\right.$ $=\mathrm{Me}), 10472,24-9 ; \mathbf{3} \mathbf{b}\left(n=12, \mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{Me}\right), 69832-85-5 ; 3 \mathbf{b}(n=$ $\left.9, \mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{Me}\right), 69832-86-6 ; 3 \mathrm{~b}\left(n=4, \mathrm{R}^{\mathrm{I}}=n-\mathrm{C}_{5} \mathrm{H}_{11}, \mathrm{R}^{2}=\mathrm{Me}\right)$, $698.32-87-7 ; 3 \mathbf{b}\left(n=2, \mathrm{R}^{1}=n-\mathrm{C}_{6} \mathrm{H}_{13}, \mathrm{R}^{2}=\mathrm{Me}\right), 69832-88-8 ; \mathbf{3} \mathbf{b}(n=$ $\left.3, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}, \mathrm{R}^{2}=\mathrm{Et}\right), 57026-68-3 ; 4 \mathrm{a}\left(n=12, \mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{Me}\right)$, 69832-89-9; 4a $\left(n=9, \mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{Me}\right), 69832-90-2 ; \mathbf{4 a}\left(n=4, \mathrm{R}^{1}=\right.$ $\left.n-\mathrm{C}_{5} \mathrm{H}_{11}, \mathrm{R}^{2}=\mathrm{Me}\right), 69832-91-3 ; 4 \mathbf{a}\left(n=3, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}, \mathrm{R}^{2}=\mathrm{Et}\right)$, 69832-92-4; 4a $\left(n=2, \mathrm{R}^{1}=n-\mathrm{C}_{6} \mathrm{H}_{13}, \mathrm{R}^{2}=\mathrm{Me}\right), 69832-93-5 ; \mathbf{4 b}(n=$ $\left.12, \mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{Me}\right), 69832-94-6 ; \mathbf{4 b}\left(n=9, \mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{Me}\right), 69832-95-7$; $4 \mathbf{b}\left(n=4, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}, \mathrm{R}^{2}=\mathrm{Me}\right), 69832-96-8 ; \mathbf{4} \mathbf{b}\left(n=3, \mathrm{R}^{1}=n-\right.$ $\left.\mathrm{C}_{5} \mathrm{H}_{11}, \mathrm{R}^{2}=\mathrm{Et}\right), 69832-97-9 ; \mathbf{4 b}\left(n=2, \mathrm{R}^{1}=n-\mathrm{C}_{6} \mathrm{H}_{13}, \mathrm{R}^{2}=\mathrm{Me}\right)$, 69832-98-0; $(E)$-5 $\left(n=12, \mathrm{R}^{1}=\mathrm{Me}\right)$, 69832-99-1; $(Z)-5\left(n=12, \mathrm{R}^{1}\right.$ $=\mathrm{Me}), 69833-00-7 ;(E)-5\left(n=9, \mathrm{R}^{1}=\mathrm{Me}\right), 69833-01-8 ;(Z)-5(\mathrm{n}=$ $\left.9, \mathrm{R}^{1}=\mathrm{Me}\right), 69833-02-9 ; 5\left(n=3, \mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}\right), 69833-03-0 ; 6$, 69833-04-1; 7, 69833-05-2; 8a, 65898-58-0; 8b, 69833-06-3; 9, 69833() $7-4$; 10, 69833-08-5; (E)-11, 58643-70-2; (Z)-11, 58643-71-3; dlmuscone, 956-82-1.

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## Reaction of 2,3,4,6-Tetramethoxybenzaldehyde with Aluminum Chloride. Selective Cleavage at Position 2 and Selective Ether Exchange at Position 3

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## Received Nowember 15, 1977

The cleavage of 2,3,4,6-tetramethoxybenzaldehyde, 1, with aluminum chloride in ether to obtain the 2-hydroxy compound, $2,{ }^{1}$ is accompanied by formation of appreciable amounts ( $38 \%$ ) of a single ethoxy-containing compound. This compound is identified as 3 . A modification for a high yield preparation of pure 2 is described.

Polymethoxybenzaldehydes with 2 -methoxy groups can cleave that group selectively with aluminum chloride. ${ }^{2-4}$ Reichstein ${ }^{2}$ formed the 2-hydroxy compound 4 from 5 using toluene as the solvent, but Robertson, ${ }^{3}$ finding that toluene cleaved all the methoxy groups, used ether as the solvent.
To monocleave 1 to 2 , we tried Robertson's method. The cleavage product had a wide melting point range after some purification and showed ethoxy peaks in ${ }^{1} \mathrm{H}$ NMR. The simplicity of the spectra suggested a mixture containing a single ethoxy compound in $38 \%$ yield. Isolation of the dimethoxyethoxyhydroxybenzaldehyde, 3, was accomplished via extraction with $5 \%$ sodium carbonate solution. Its identity was established by methylation and comparison with known ethoxytrimethoxybenzaldehydes. The ethoxy should be either in the 3 position (next to the phenol) or the 6 position (next to the formyl group). Accordingly, 3 -ethoxy-2,4,6-trimethoxyand 6 -ethoxy-2,3,4-trimethoxybenzaldehydes, 6 and 7 , were prepared respectively from the phenols 8 and 9 via ethylation to 10 and 11 and formylation to 6 and 7 . By increasing the


|  | $\mathrm{R}_{1}$ | $\mathrm{R}_{2}$ | $\mathrm{R}_{3}$ | $\mathrm{R}_{4}$ |
| ---: | :--- | :--- | :--- | :--- |
| 1 | CHO | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3} \mathrm{O}$ |
| $\mathbf{2}$ | CHO | H | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3} \mathrm{O}$ |
| 3 | CHO | H | $\mathrm{CH}_{3} \mathrm{CH}_{2}$ | $\mathrm{CH}_{3} \mathrm{O}$ |
| 4 | CHO | H | $\mathrm{CH}_{3}$ | H |
| 5 | CHO | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | H |
| 6 | CHO | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3} \mathrm{CH}_{2}$ | $\mathrm{CH}_{3} \mathrm{O}$ |
| 7 | CHO | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}$ |
| 8 | H | $\mathrm{CH}_{3}$ | H | $\mathrm{CH}_{3} \mathrm{O}$ |
| 9 | H | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | $\mathrm{OH}_{3}$ |
| 10 | H | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3} \mathrm{CH}_{2}$ | $\mathrm{CH}_{3} \mathrm{O}$ |
| 11 | H | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3} \mathrm{CH} \mathrm{O}$ |
| 12 | CHO | H | H | $\mathrm{CH}_{3} \mathrm{O}$ |
| 13 | CHO |  | $-\mathrm{CH}_{2}-$ | $\mathrm{CH}_{3} \mathrm{O}$ |

